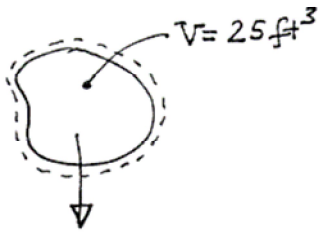


PROBLEM 1.8



$$F_{\text{grav}} = 3.5 \text{ lbf}$$

$$g_{\text{moon}} = 5.47 \text{ ft/s}^2$$

$$g_{\text{mars}} = 12.86 \text{ ft/s}^2$$

Accordingly

$$(F_{\text{grav}})_{\text{mars}} = \left(\frac{g_{\text{mars}}}{g_{\text{moon}}} \right) (F_{\text{grav}})_{\text{moon}}$$

$$= \left(\frac{12.86 \text{ ft/s}^2}{5.47 \text{ ft/s}^2} \right) (3.5 \text{ lbf}) = 8.23 \text{ lbf} \quad \leftarrow F_{\text{grav, mars}}$$

The density is $\rho = m/V$. Applying Eq. (*) with data on mars

$$m = \left(\frac{8.23 \text{ lbf}}{12.86 \text{ ft/s}^2} \right) \left| \frac{\overset{\text{rounded}}{32.2 \text{ lb} \cdot \text{ft/s}^2}}{1 \text{ lbf}} \right| = 20.61 \text{ lb}$$

Then

$$\rho = \frac{20.61 \text{ lb}}{25 \text{ ft}^3} = 0.824 \frac{\text{lb}}{\text{ft}^3} \quad \leftarrow \rho$$

PROBLEM 1.9

$$F = ma = m(60g) = 60mg$$

$$= 60(5016)\left(32.2 \frac{\text{ft}}{\text{s}^2}\right) \left| \frac{1 \text{ lbf}}{32.2 \text{ lb}\cdot\text{ft}/\text{s}^2} \right| = 3000 \text{ lbf} \quad \leftarrow F$$

↖ rounded

PROBLEM 1.10.

Eq. 1.8 is used in both parts: $n = m/M$, where M is from Tables A-1.

(a) $m = M n$, $n = 10 \text{ kmol}$

Air: $m = (28.97 \text{ kg/kmol})(10 \text{ kmol}) = 289.7 \text{ kg}$

H₂O: $m = (18.02 \text{ kg/kmol})(10 \text{ kmol}) = 180.2 \text{ kg}$

Cu: $m = (63.54 \text{ kg/kmol})(10 \text{ kmol}) = 635.4 \text{ kg}$

SO₂: $m = (64.06 \text{ kg/kmol})(10 \text{ kmol}) = 640.6 \text{ kg}$

(b) $n = m/M$, $m = 20 \text{ lb}$

Air: $n = (20 \text{ lb}) / (39.94 \text{ lb/lbmol}) = 0.501 \text{ lbmol}$

H₂: $n = (20 \text{ lb}) / (2.016 \text{ lb/lbmol}) = 9.921 \text{ lbmol}$

N₂: $n = (20 \text{ lb}) / (28.01 \text{ lb/lbmol}) = 0.714 \text{ lbmol}$

C: $n = (20 \text{ lb}) / (12.01 \text{ lb/lbmol}) = 1.665 \text{ lbmol}$